CHE680 Advanced Analytical Chemistry



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<u>Lecture</u> 4:30 PM – 5:45 PM, MW, SAMC201

Office Hour 11:00 AM- 11:50 AM MWF or by Appointment

Course Description

- 1. Basic theory of analytical chemistry (concentration, balancing, stoichiometry, acid-base, titration, oxidationreduction, calibration, etc) will be reviewed.
- 2. Principles and applications of various analytical techniques used in modern analytical laboratory will be introduced.
- 3. These techniques include FTIR, UV/Vis, Raman, XPS, AES, AAS, GC, HPLC, Mass, and so on.
- 4. Advantages and limitations of individual analytical technique for quantitative and qualitative chemical analysis are also discussed.
- 5. Selected applications of these techniques will be demonstrated using either unknown or known samples.

Course Evaluation

Exam 1 (10/23/19)	400	40.0%
Exam 2 (12/4/19)	450	45.0%
HW	150	15.0%
TOTAL	1000	100.0%
93–100% A;	90-92.9% A-;	
87-89.9% B+;	83-86.9% B;	80-82.9% B-;
77 – 79.9% C+;	71 – 76.9% C;	68 – 70.9% C-

Why Analytical Chemistry is Important?

Chemical analysis is important to various fields including health, forensic, pharmaceutical, environments, food, national security safety, energy, etc.



pharmaceuticals



forensics

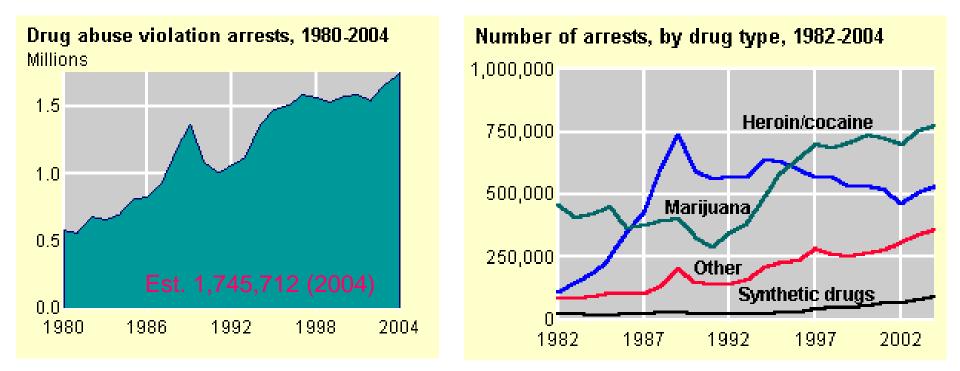


health



national security

e.g. Drug-Related Crime Statistics



Federal drug control spending: \$8,179M (FY 1988) and \$11,679M (FY 2004)

About 10,000 new job within 10 years estimated by the American Academy of Forensic Sciences (AAFS)

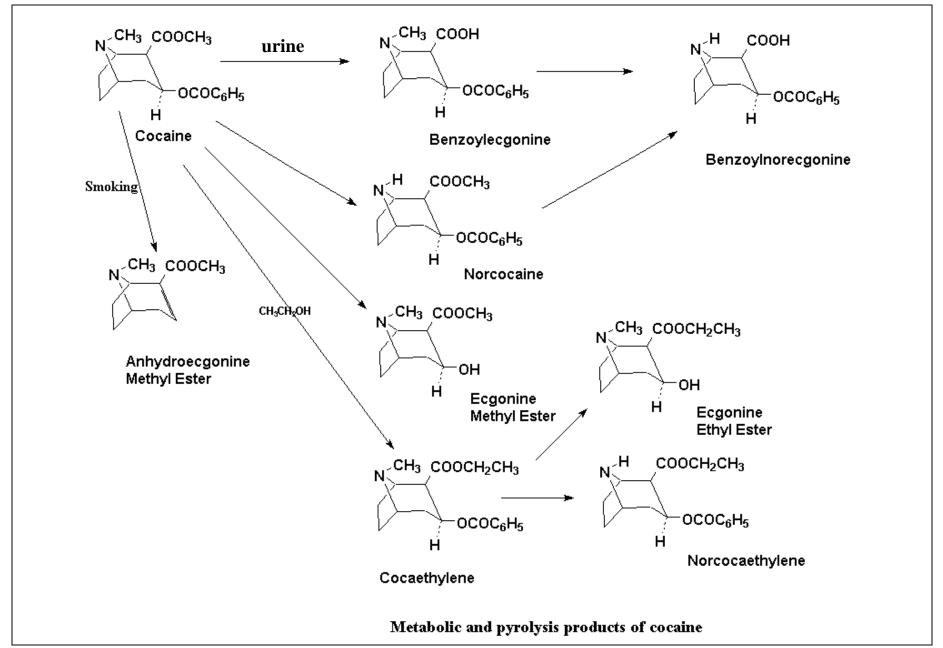
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	New York State Police Crime Lab System	Location Albany, NY United States Of America		
Application Deadline	8/11/2017			
		Responsibilities		
Responsible for the adr of CoDIS finds.	ninistration of the technical and scientific operations of	f the CoDIS computer network within the DNA Data Bank, including supervision of procedures and reports		
Ensures that the DNA I Forensic DNA Data Ba		the New York State guidelines for DNA analysis, and that all protocols meet professional standards for		
Oversees development	s within the fie			
View	Forensic Scientist - Drug Chemistry			
Organization Name	New York State Police Crime Lab System	Location Port Crane, NY United States Of America		
Application Deadline				
		Responsibilities		
Vill be familiar with and Vill successfully compl ompetency and/or pro	d aware of work completed and analyses performed in ete drug chemistry training under the guidance of the ficiency tests will be completed before assuming any	other laboratory sections. Supervisor of Forensic Services - Technical Coordinator - Drug Chemistry. Appropriate training with		
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state, and local law enforcement agencies. The following encompasses the facets of these administrative duties related to a discipline/sub-discipline: in order to maintain chain of

Cocaine and Metabolites



Application of Chemistry to Forensic Investigations: Drug Test

a. in the laboratory

samples (urine, blood, hair, etc) gas chromatographymass spectrometry (GC-MS)

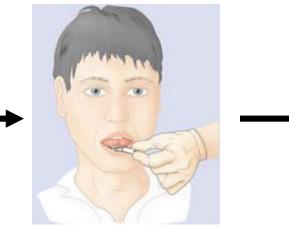
qualitative and quantitative analysis



b. on site

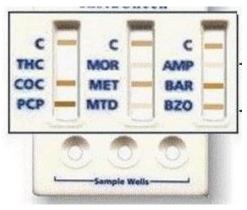


drug test kit



saliva, urine, sweat, hair, nail, or blood test

color change



protein and drug complex formation on the surface



 Qualitative analysis: the determination of <u>identities of</u> <u>elements</u> and <u>compounds</u> that are present in a sample.

 Quantitative analysis: the determination of <u>the amount</u> (concentrations) of each element in a sample.



What are in it? : Sugar, caffeine, theobromine, etc How much? : sugar: 200 mg, caffeine: 35 mg, etc

Tools in Chemical Analysis

- 1. Spectroscopy (XPS, Auger, AA/AE, UV/Vis, IR, Raman, NMR, ESR, Mössbauer, etc)
- 2. Chromatography (GC, HPLC, IC, TLC, SFC, SEC, CE, etc)
- 3. Microscopy (SEM, TEM, AFM, FM, etc)
- 4. Diffraction (X-ray, Neutron, Powder, Single crystal, etc)
- 4. Mass spectrometry (GC/MS, LC/MS)
- 5. Light Scattering
- 6. Thermal analysis (DSC, TGA, DTA, etc)
- 7. Electroanalytical Methods (Potentiometric, Coulometric, Voltammetric, and Capacitance Measurements, etc)
- 8. SPR (Surface Plasmon Resonance), Ellipsometry, X-ray (Neutron) Reflectivity
- 9. QCM (Quartz Crystal microbalance)
- 10. CD (Circular Dichroism)
- 11. Neutron Activation Analysis
- 12. Magnetic Susceptibility



No single technique is perfect and many complementary tools are required

Factors To Be Considered

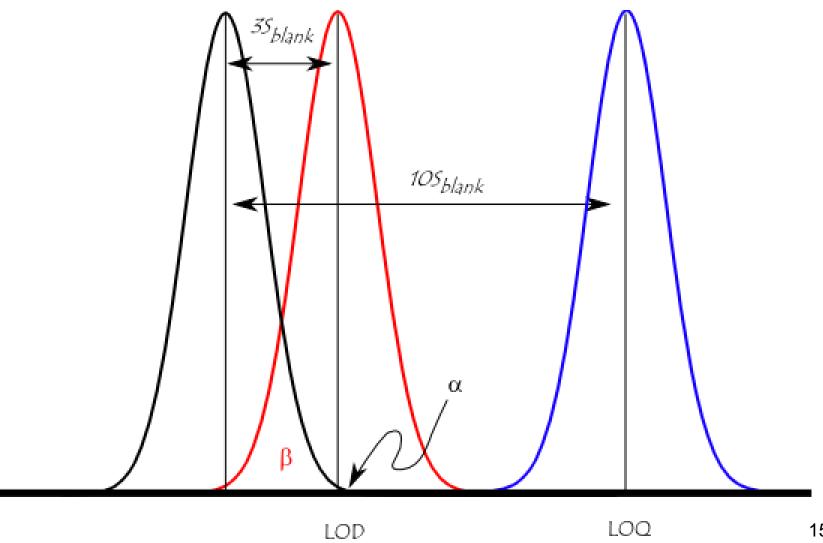
- 1. Sensitivity and selectivity
 - Sensitivity (LOD, LOQ)
 - Selectivity
- 2. Size of sample and states (air, liquid, or solid)Destructive analysis
 - Destructive analysis
 Nondestructive analysis
- 3. Analysis time
- 4. Availability and cost
- 5. Feasibility and convenience
- 6. Safety

LOD and LOQ

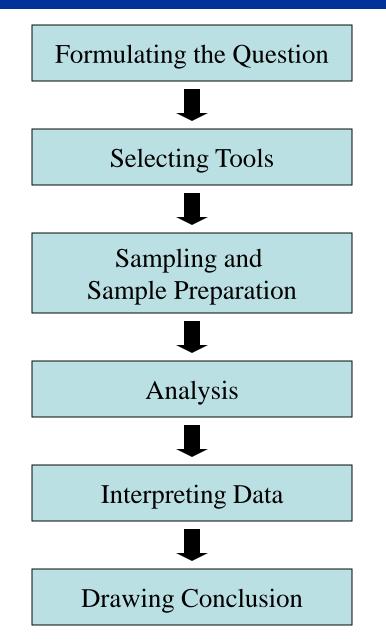
• LOD (Limit of Detection): the lowest quantity of a substance that can be distinguished from the absence of that substance (a *blank value*) within a stated confidence limit (~3 times of blank std).

• LOQ (Limit of Quantification): the limit at which we can reasonably tell the difference between two different values (~10 times of blank std).

LOD and LOQ



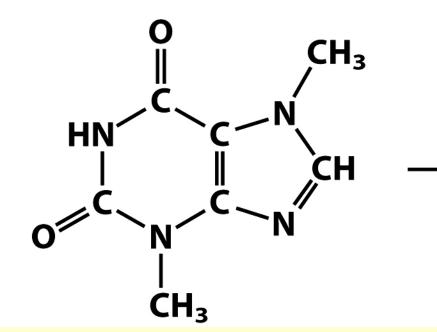
General Steps in Chemical Analysis

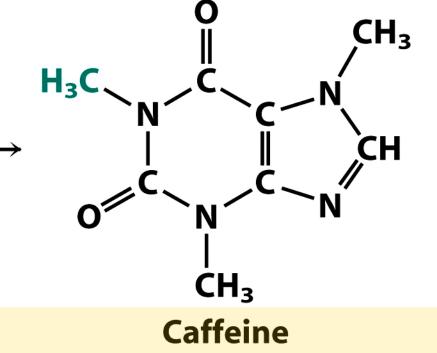


Problem: How much caffeine in a chocolate bar?



Target Compounds: Caffeine and its Precursor



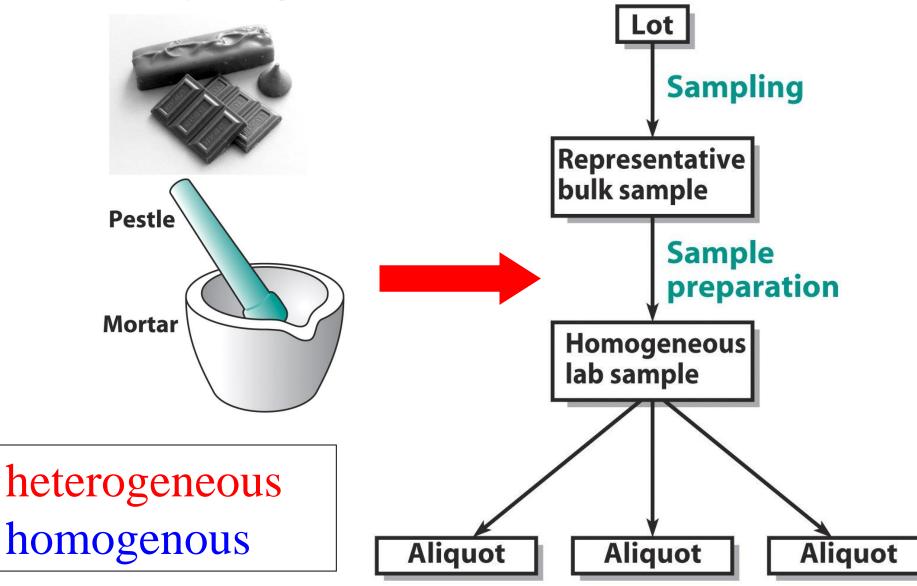


Theobromine Diuretic, smooth muscle relaxant, cardiac stimulant, and vasodilator

Central nervous system stimulant and diuretic

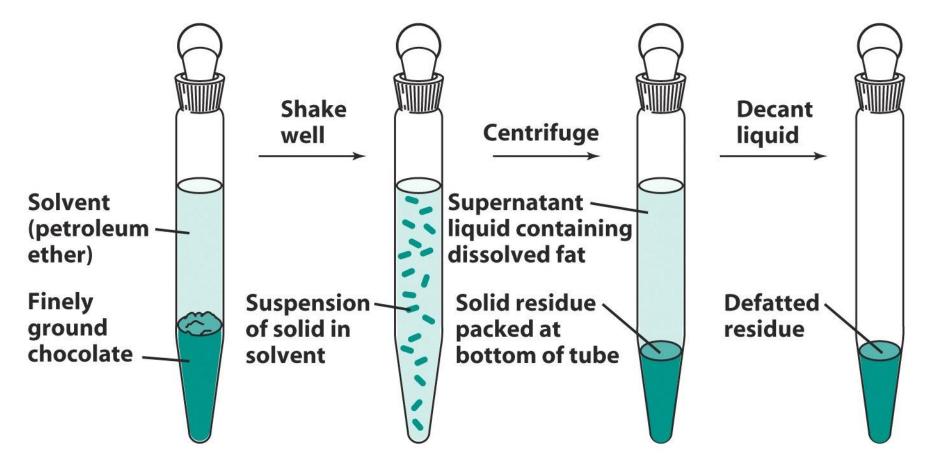
Sampling and Sample Preparation

Heterogeneous sample



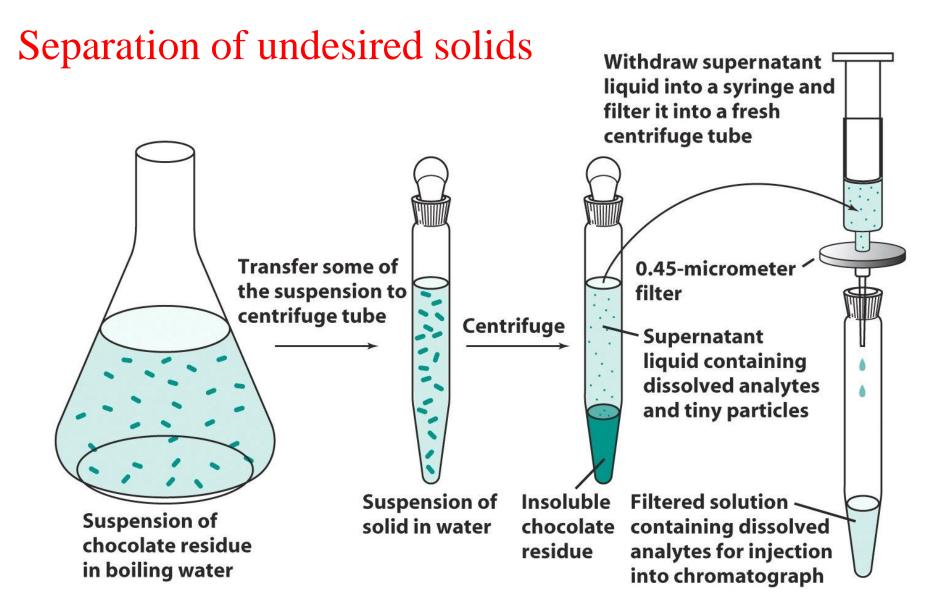
Sample Preparation

Separation of interference (fats)

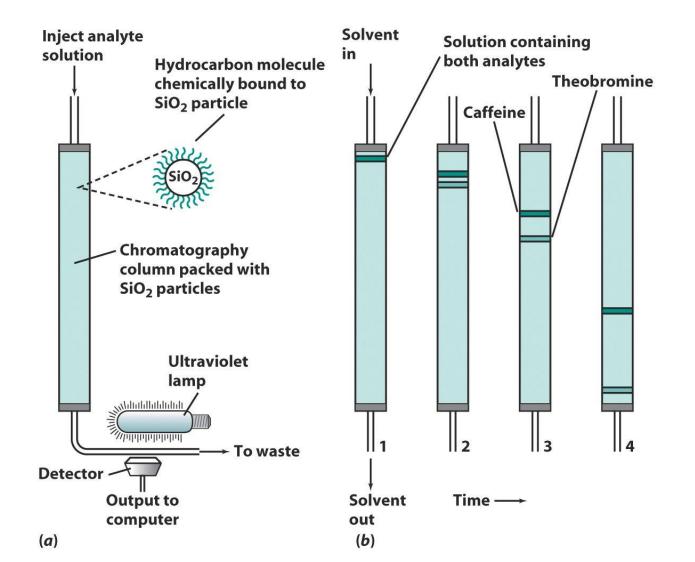


Find good solvents for fats, but poor solvent for caffeine

Sample Preparation



Analysis of Caffeine and Theobromine by Liquid Chromatography



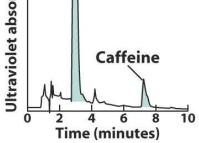
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Analytical Results of Caffeine and Theobromine by Liquid Chromatography

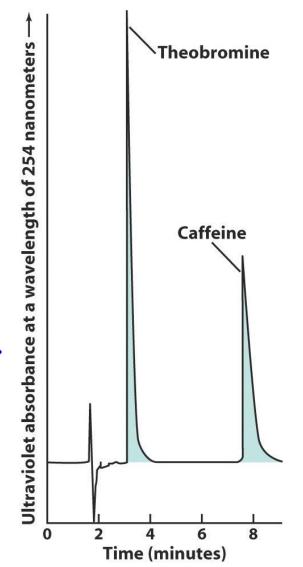
Results from chocolate bar indicating existence of two target compounds without knowing their concentrations (qualitative analysis)

We need standard for the determination

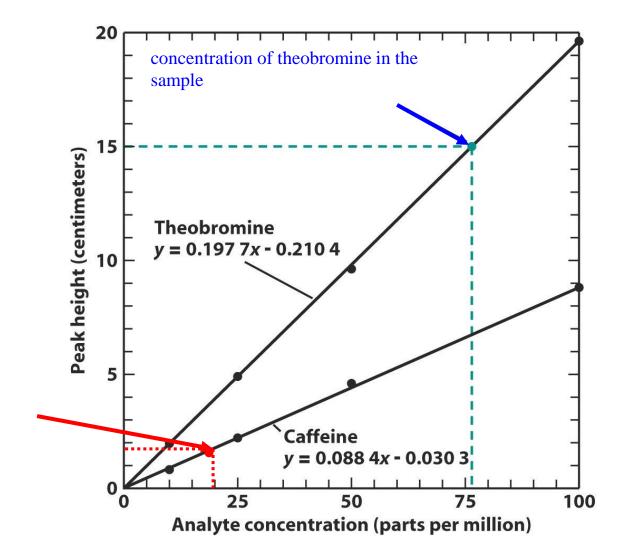
We need standard for the determination of concentrations



Results from one standard sample with known caffeine and theobromine concentrations.



Construction of Calibration Curve from a Series of Standard Samples

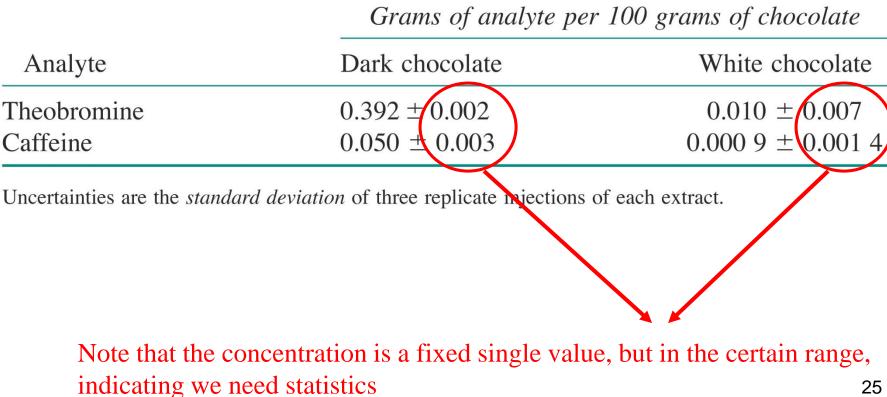


concentration of caffeine in the sample

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Analytical Results

TABLE 0-1 Analyses of dark and white chocolate



Caffeine Content of Beverage and Foods

Caffeine content of beverages and foods

Source	Caffeine (milligrams per serving)	Serving size ^a (ounces)
Regular coffee	106–164	5
Decaffeinated coffee	2–5	5
Tea	21-50	5
Cocoa beverage	2-8	6
Baking chocolate	35	1
Sweet chocolate	20	1
Milk chocolate	6	1
Caffeinated soft drinks	36–57	12

a. 1 ounce = 28.35 grams.

TABLE 0-2

SOURCE: Tea Association (http://www.chinamist.com/caffeine.htm).

Question is How?

Sample preparation: wet chemistry

Standard sample preparation: wet chemistry

≻Chemical analysis: Instrumental analysis

➢Data analysis