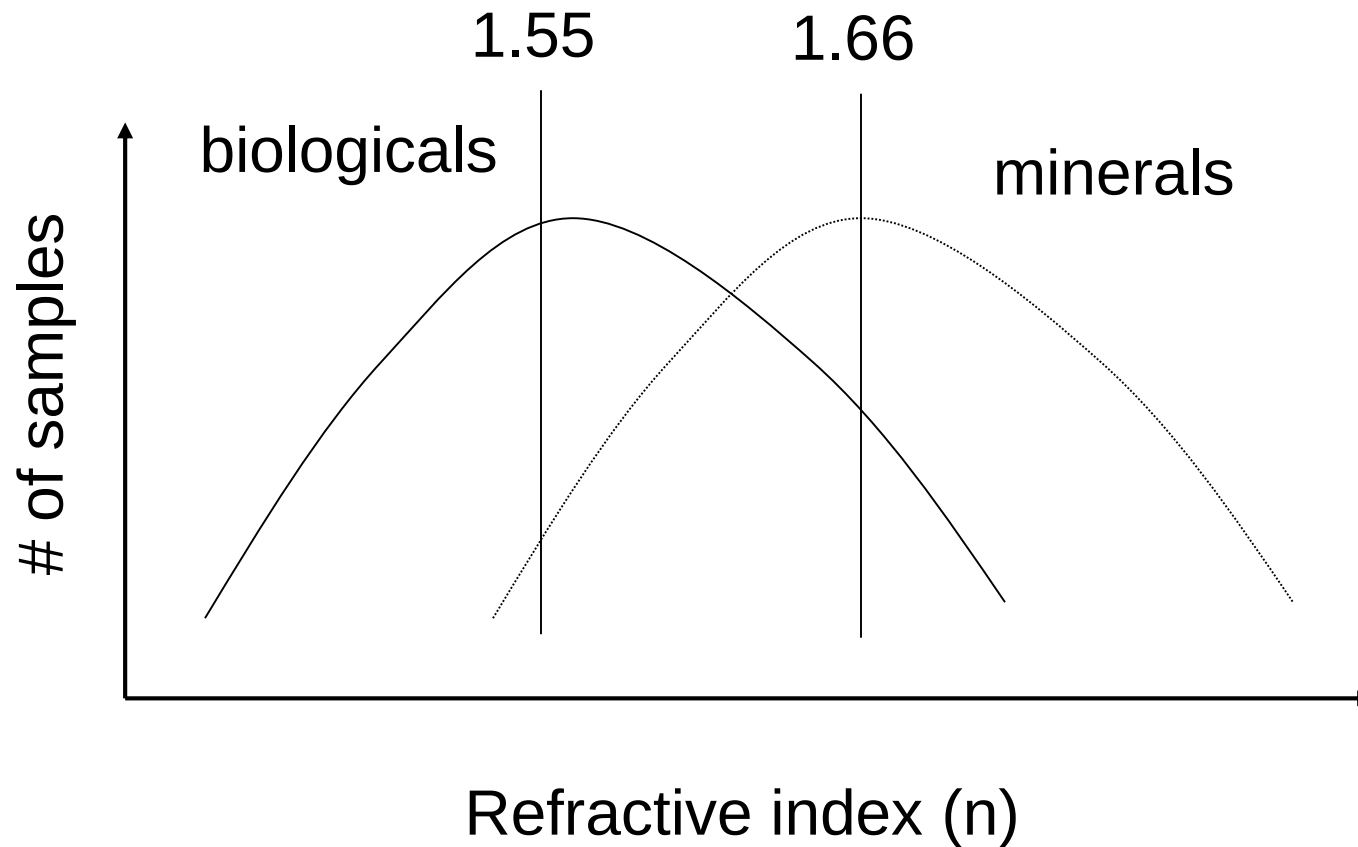


Lecture 8

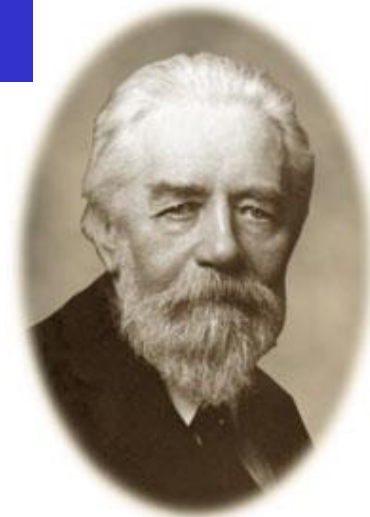
Measuring Refractive Index

Refractive Index

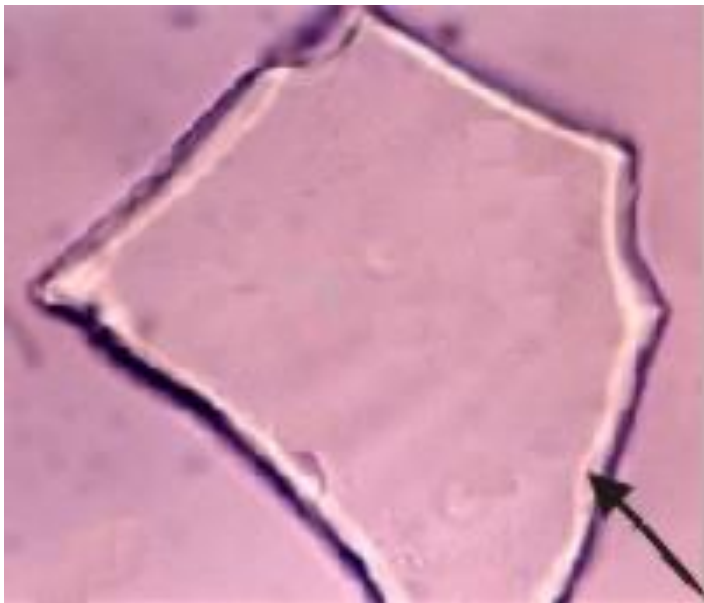


Becke Line Test

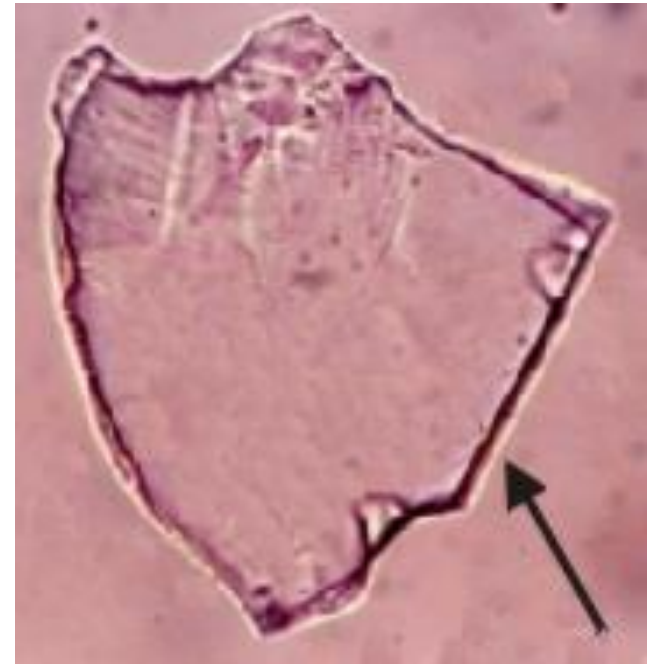
1. The indices of refraction of the particle and surrounding medium are different.
2. The microscope is defocused (on the sample).



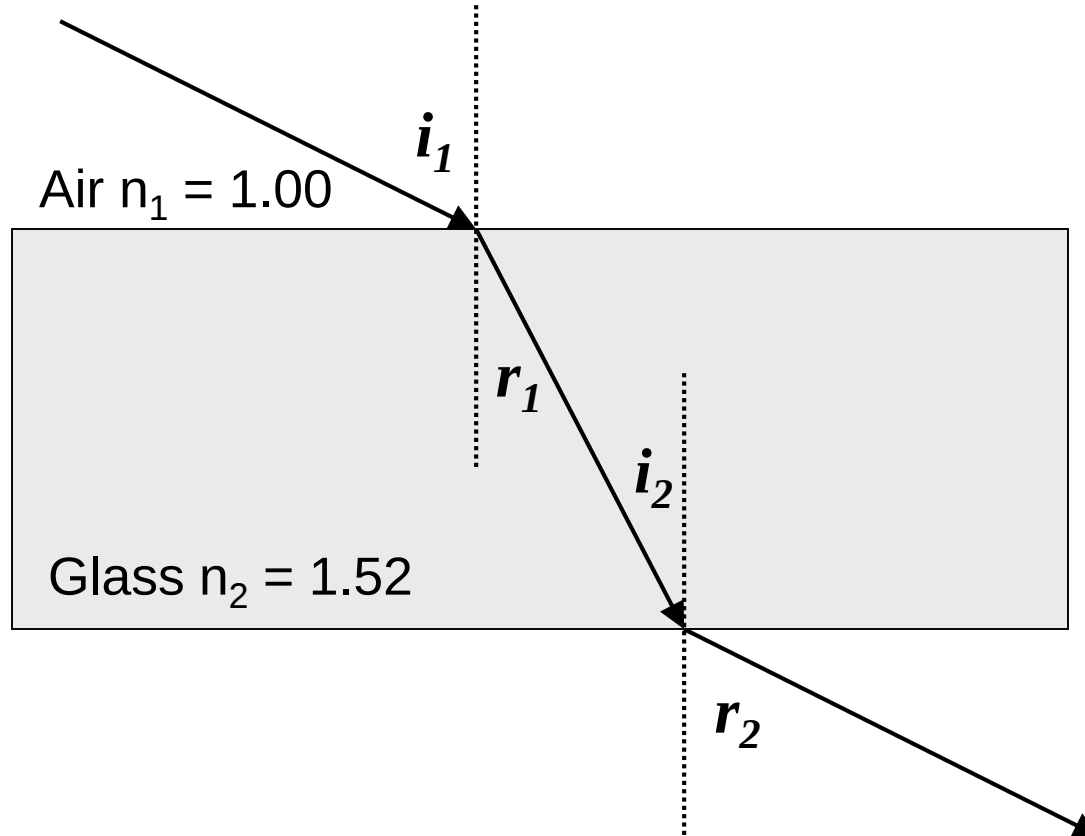
Friedrich Becke
(1855-1931)



Becke line

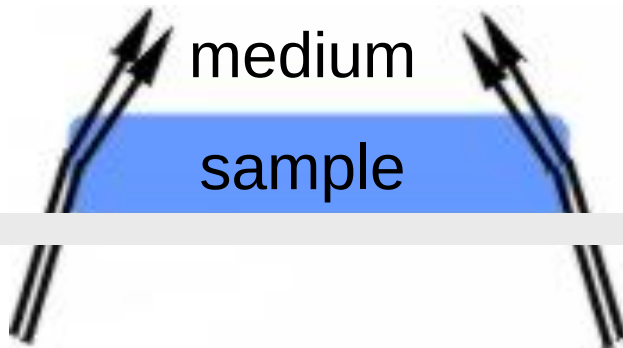
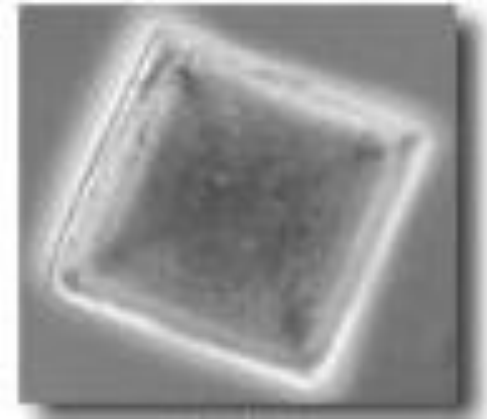


Refraction of Light

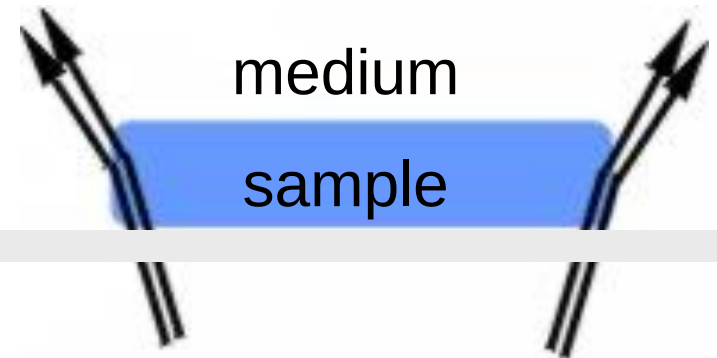


Light bends toward (or away) from surface normal.

Formation of Becke Line



$$n_{\text{medium}} < n_{\text{sample}}$$



$$n_{\text{medium}} > n_{\text{sample}}$$

Can be used to find refractive index of unknown materials using standard liquids

How to Check Becke Line

1. Prepare a series of samples in media with different refractive indices
2. High contrast > close down condenser aperture
3. Control wavelength > yellow is best > use orange filter
4. Raise the focus > lower the stage > rotate toward you
5. Decide the refractive index depending on the position of Becke line

Refractive Index Measurements via Becke Line Test

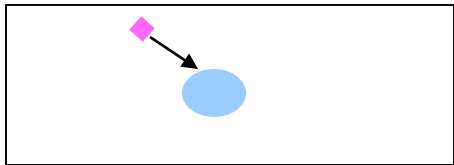
1. One unknown sample
2. Choose standard refractive index liquids (1.4 and 1.8)
3. Decide whether refractive index is equal, greater, or less than refractive index you choose

Recrystallization of Inorganic/Organic Salts

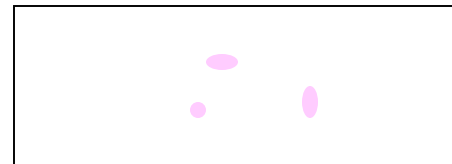
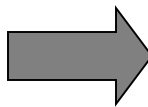
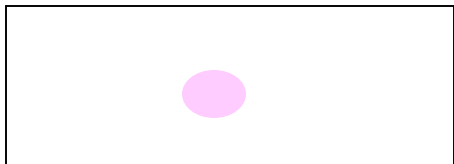
1. Sublimation

2. Evaporation

- 10X Objective
- Supersaturation



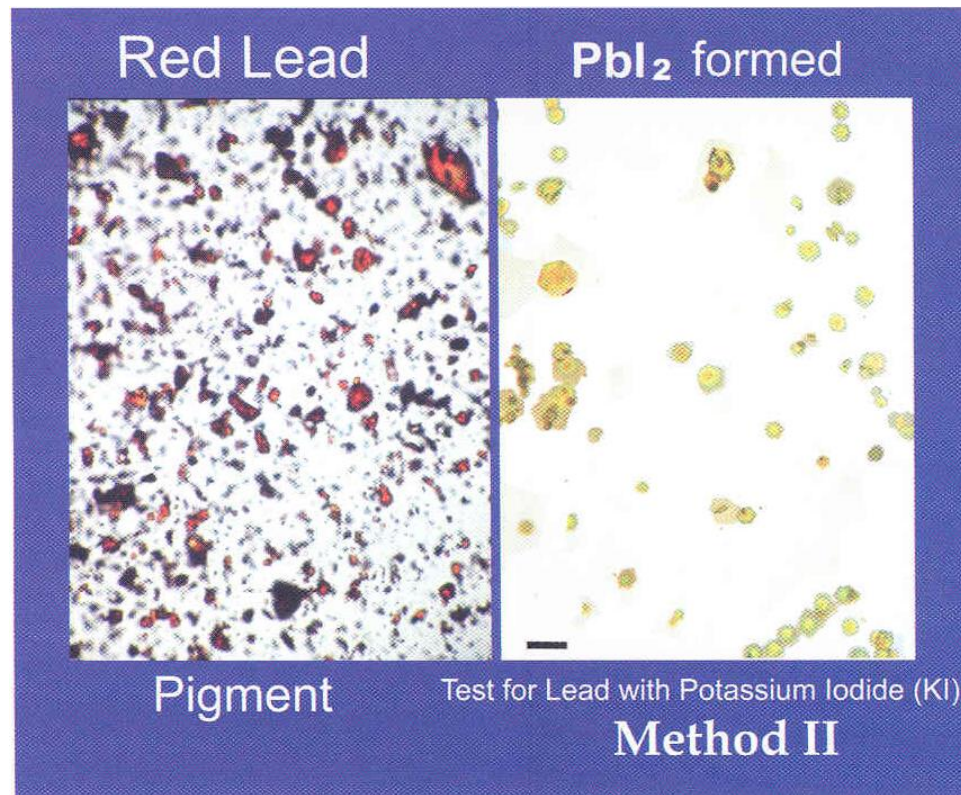
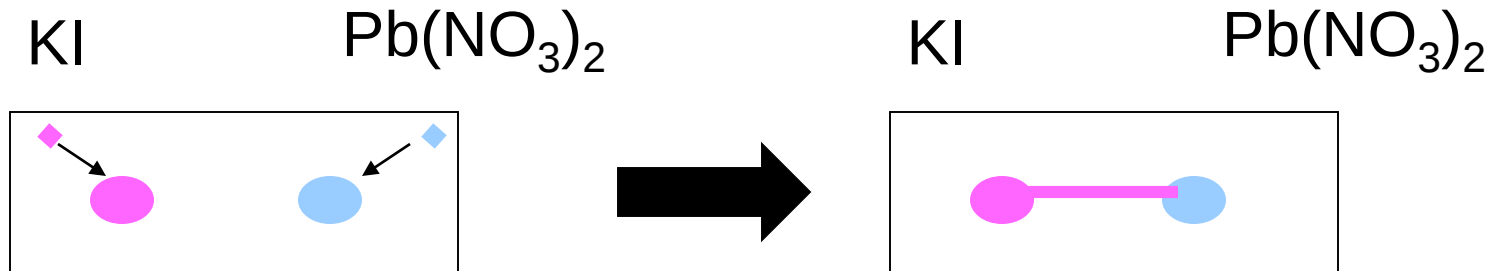
3. Transfer saturated solution “Huffing”



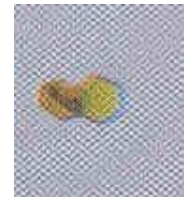
Melt-Recrystallization Sample Preparation



Detection of Lead



PbI_2



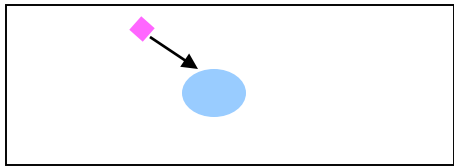
Exercise 5: Fiber Sample Preparation (see handout)

1. Three samples (hair and two fibers)
2. Your hair (cut one or two hair of $\sim 1''$)
3. Choose your sample in the list
4. Label your samples with marker (initial and sample name) on the slide glass
5. Cure them for a week
6. Place them in your assigned positions in a sample box by next week

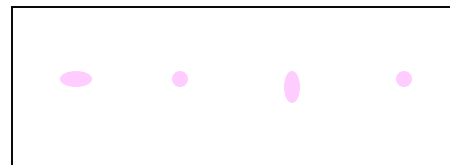
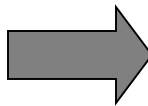
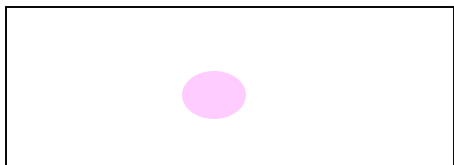
Exercise 6: Mineral Sample Preparation (see handout)

1. DO NOT contaminate your sample
2. Use very small portion of your sample

3. Evaporation



4. Transfer saturated solution “Huffing”



Lecture 9

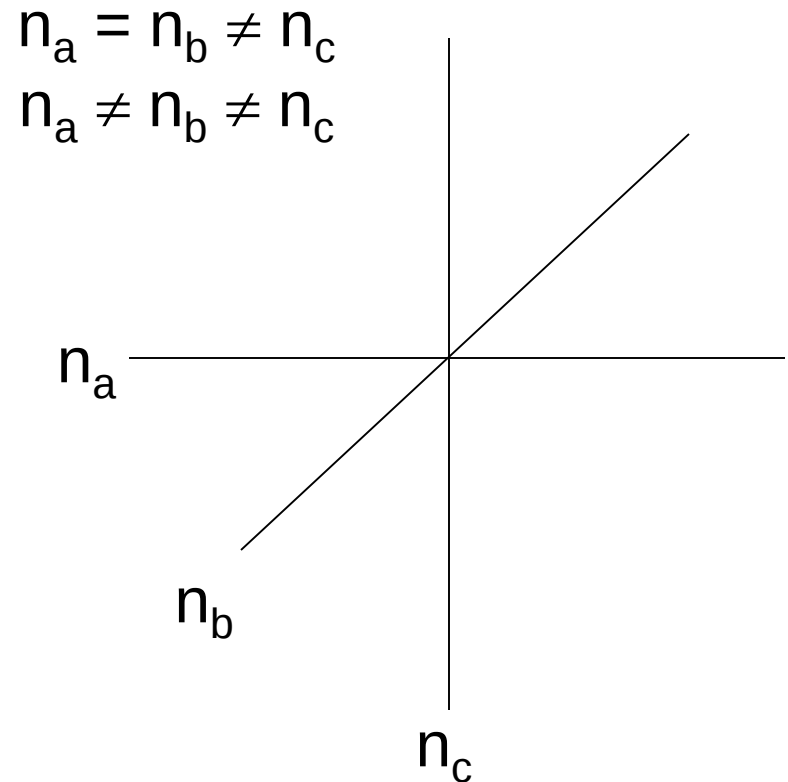
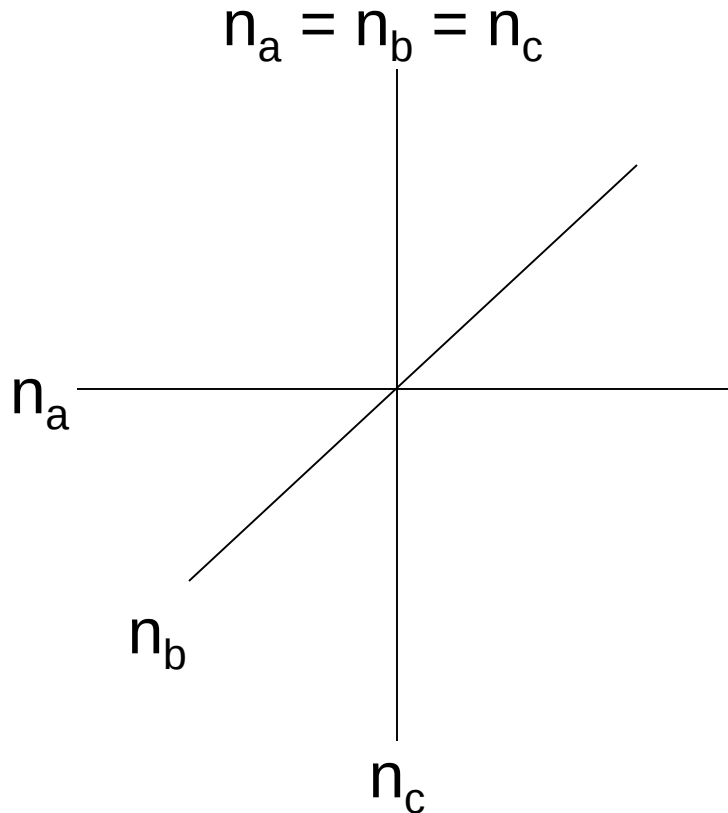
Crystal Optics

Isotropic vs Anisotropic

Refractive index: interaction between electromagnetic light and materials (ions and molecules)

Isotropic: Equal refractive index in all directions (one n)

Anisotropic: Not equal refractive index in all directions (two or three n)



Cubic (Isotropic) Indicatrix

- Same refractive index in all directions
- Light travelling in all directions has same speed
- Cross sections are always circle (radius n)
- Cubic crystal system only

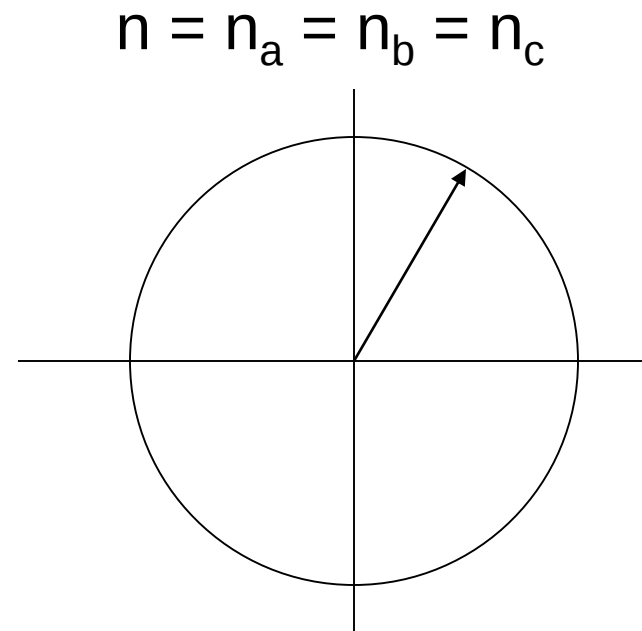
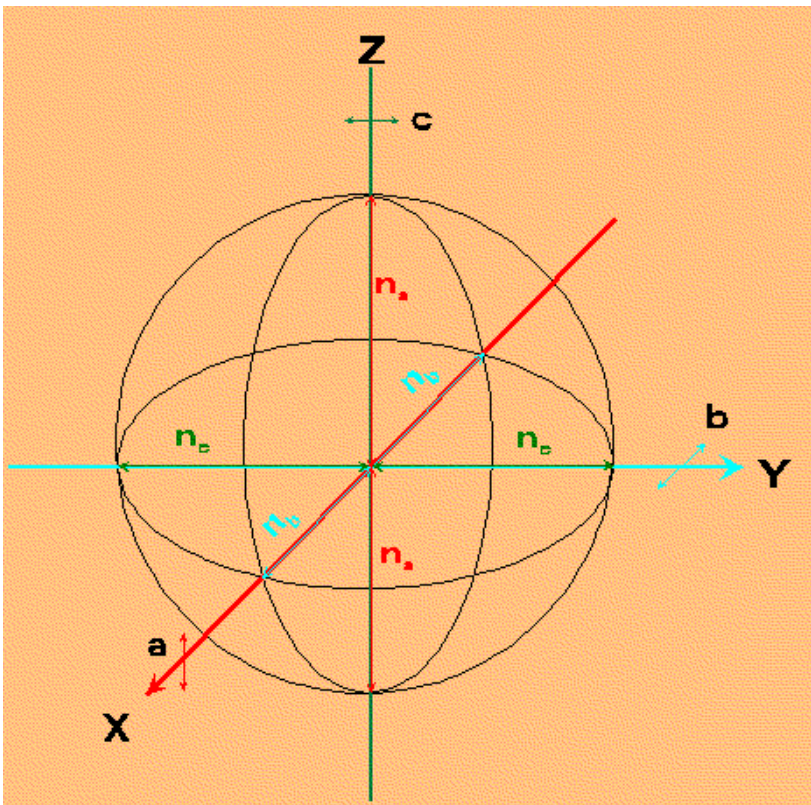
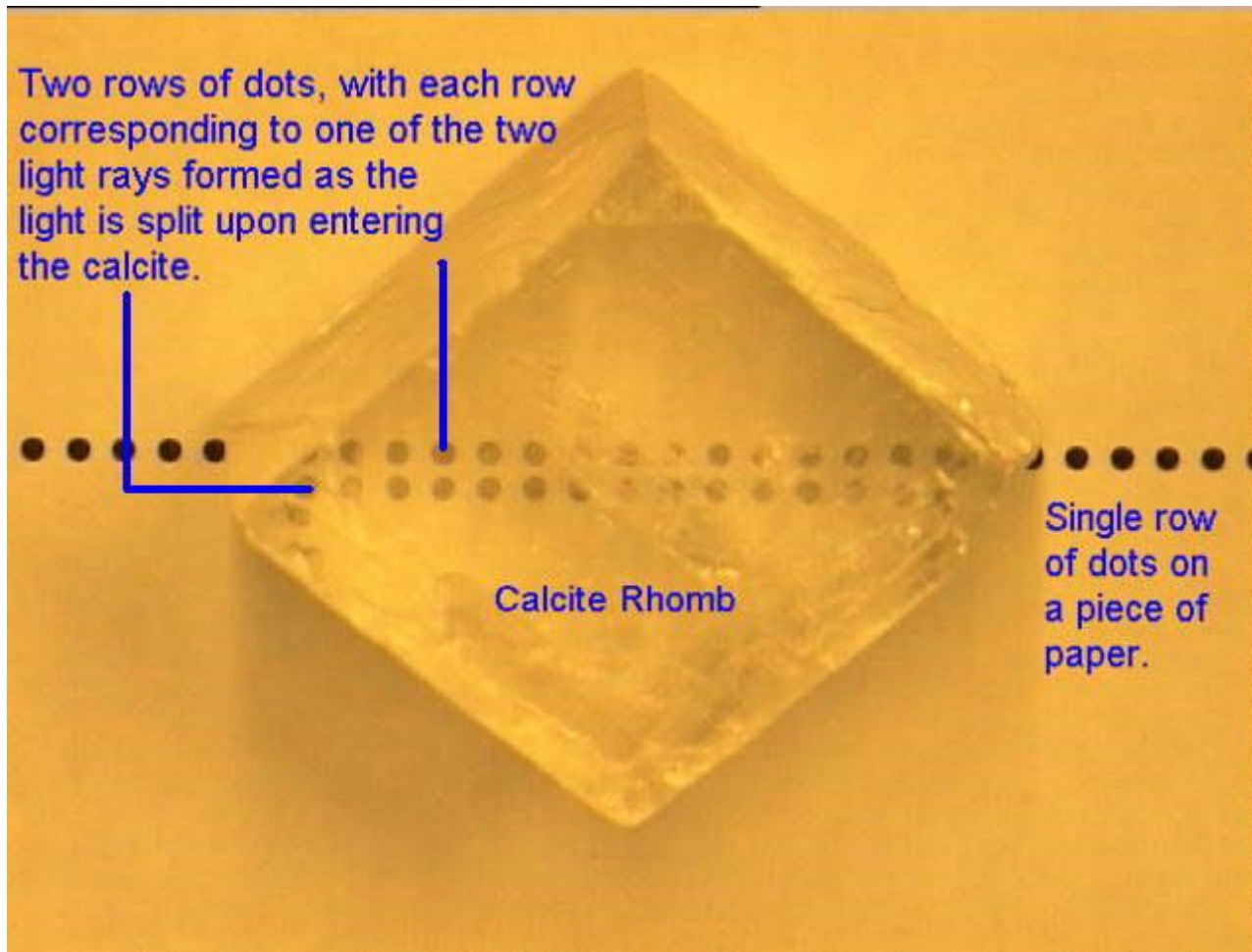


Image through Calcite



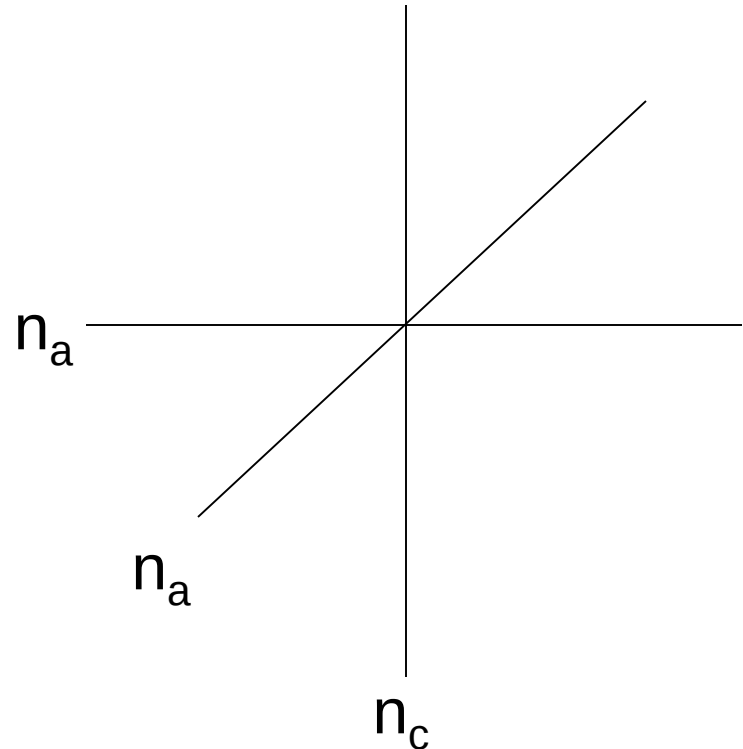
Uniaxial Indicatrix 1

- Two refractive indices
- $n_a = n_b (\omega) \neq n_c (\varepsilon)$
- Light travels in different speed depending on directions
- Hexagonal and tetragonal crystal systems

$$n_a = n_b (\omega) \neq n_c (\varepsilon)$$

$$\omega < \varepsilon \text{ or}$$

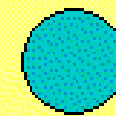
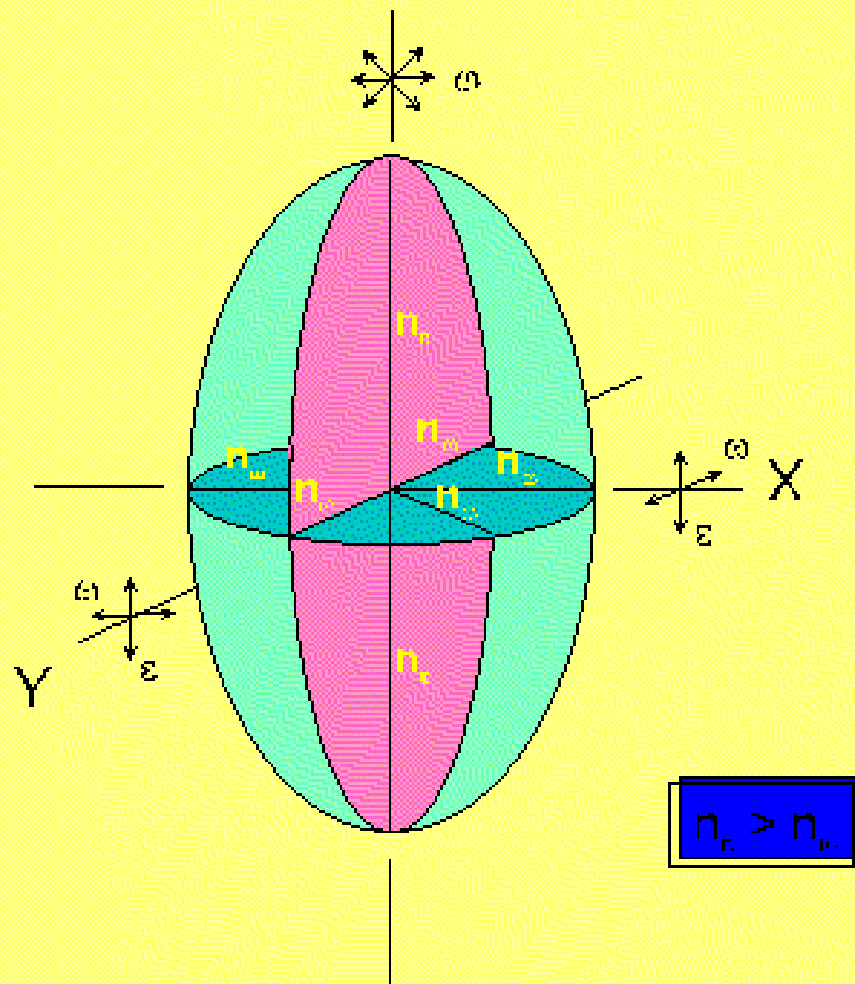
$$\omega > \varepsilon$$



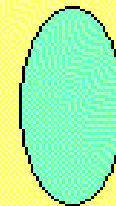
Uniaxial Indicatrix (Positive)

If $\omega < \varepsilon$ (positive)

Z = optic axis = c-axis



Circular Section
Radius = n_ω



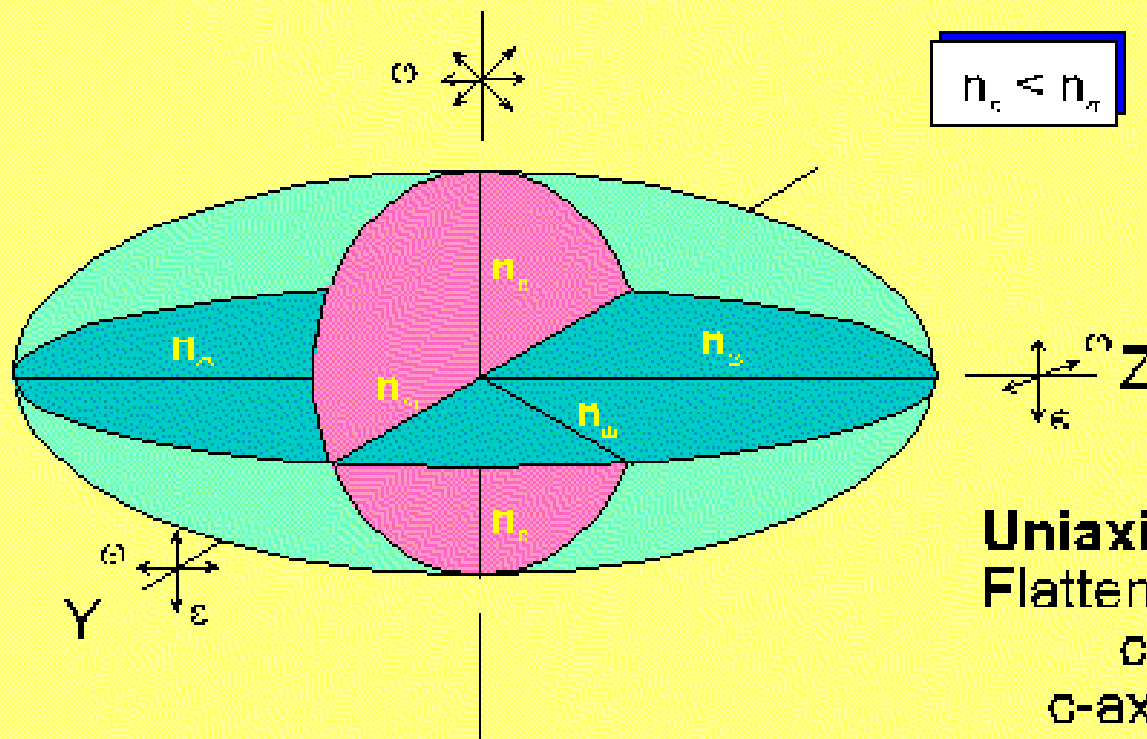
Principle Sections
Radii = n_x and n_ω

Uniaxial Positive Indicatrix
Elongated along the optic axis
c-axis = optic axis
c-axis = Z Indicatrix axis

Uniaxial Indicatrix (Negative)

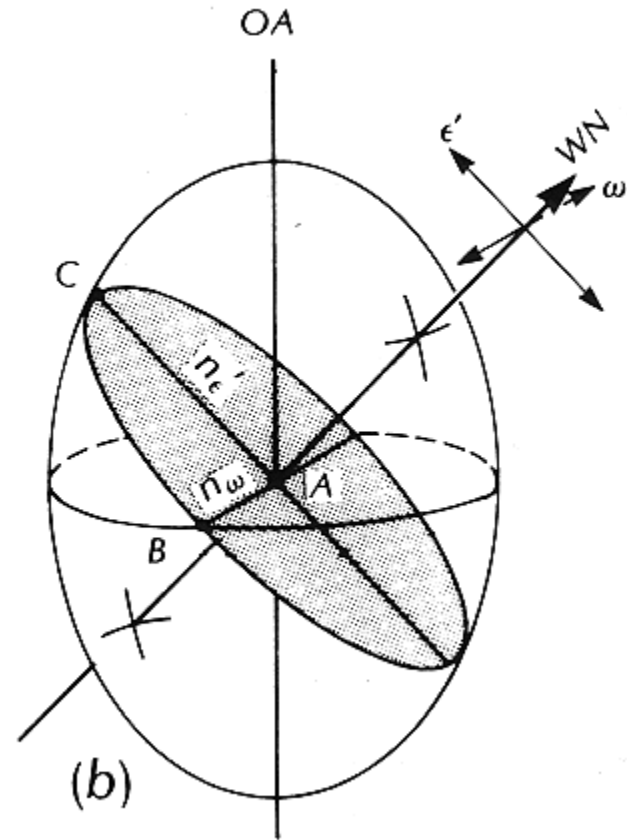
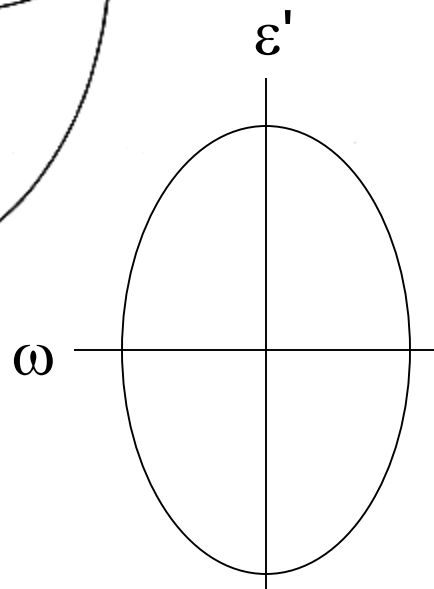
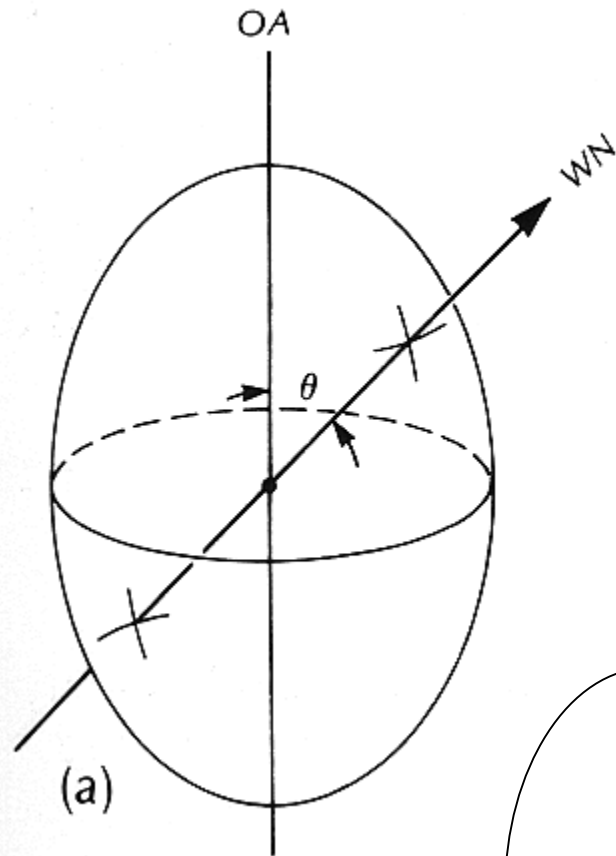
If $\omega > \varepsilon$ (negative)

X = optic axis = c-axis



Uniaxial Negative Indicatrix
 Flattened along the optic axis
 c-axis = optic axis
 c-axis = X indicatrix axis

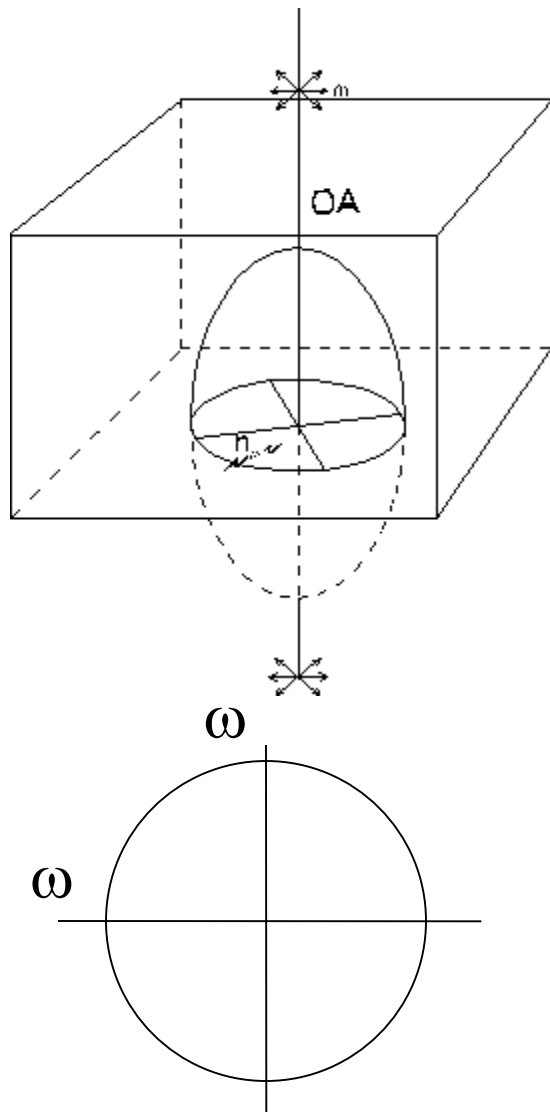
Random Section Vibration Directions



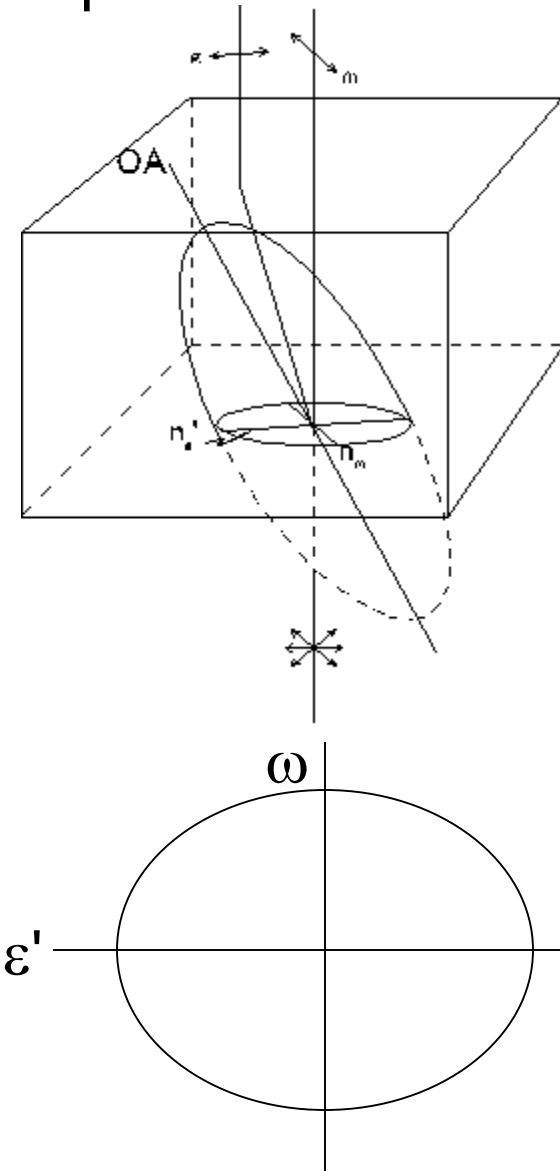
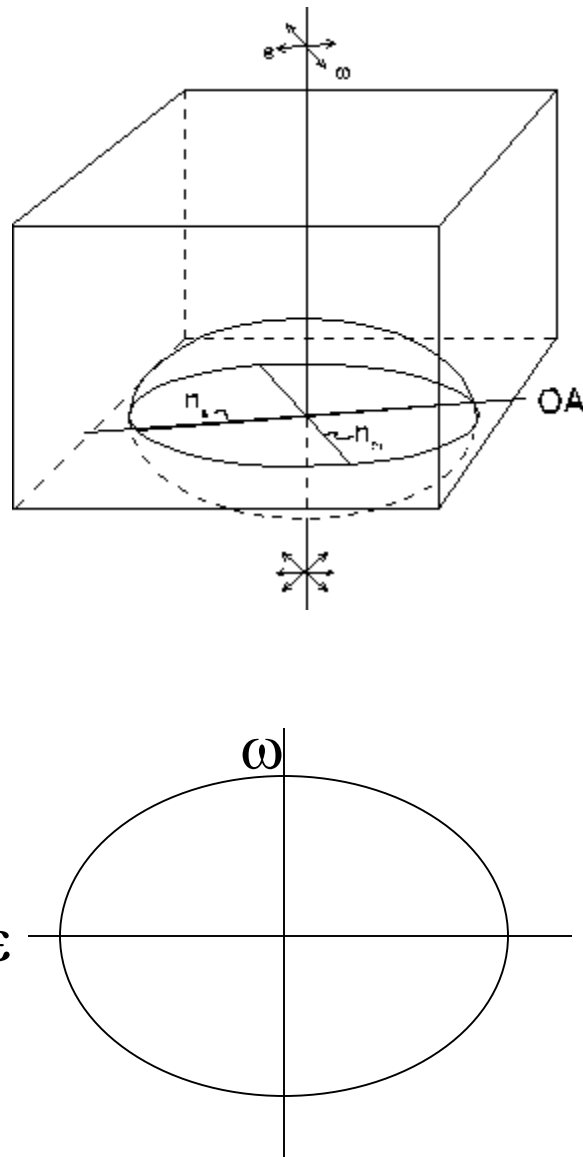
$$\omega < \epsilon' < \epsilon$$

Uniaxial Crystal: Orientation-Dependence

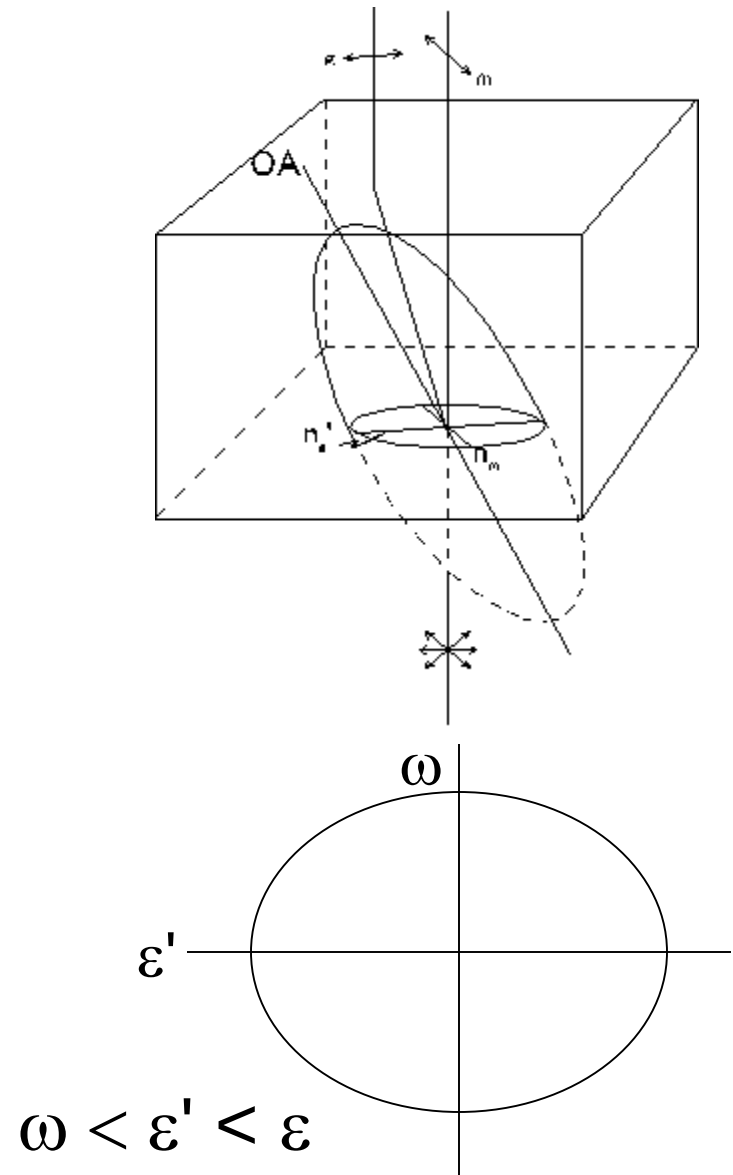
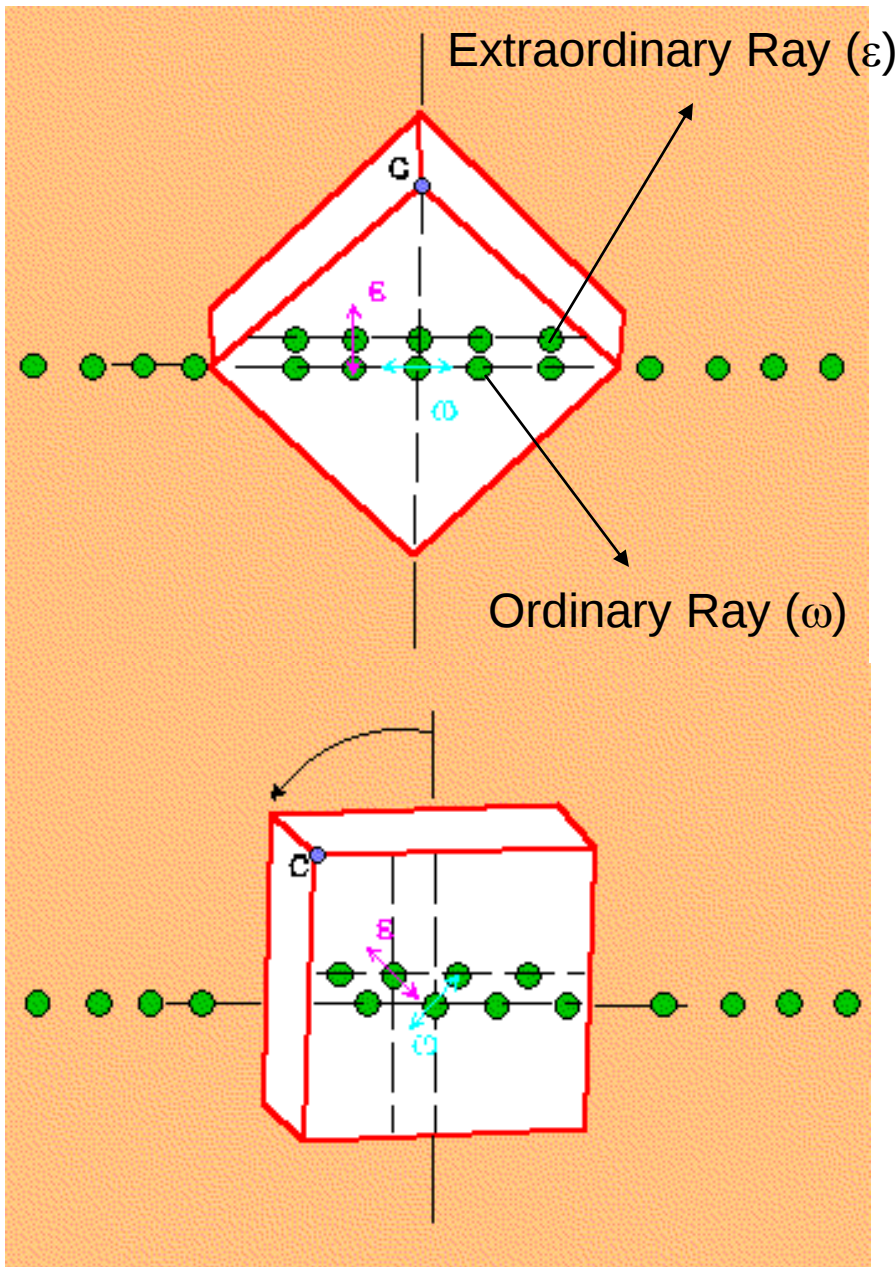
isotropic



anisotropic



Calcite Double Refraction



Birefringence: Double Refraction

Material with two (or more) refractive indices

For calcite,

$n_{\text{omega}} = 1.658$ (parallel to c axis, ordinary ray, regardless of the direction).

$n_{\text{epsilon}} = 1.486$ to 1.658 (perpendicular to c axis extraordinary ray, dependant on the direction)

calcite, $\Delta n = 0.172$ (two images with very large separation)

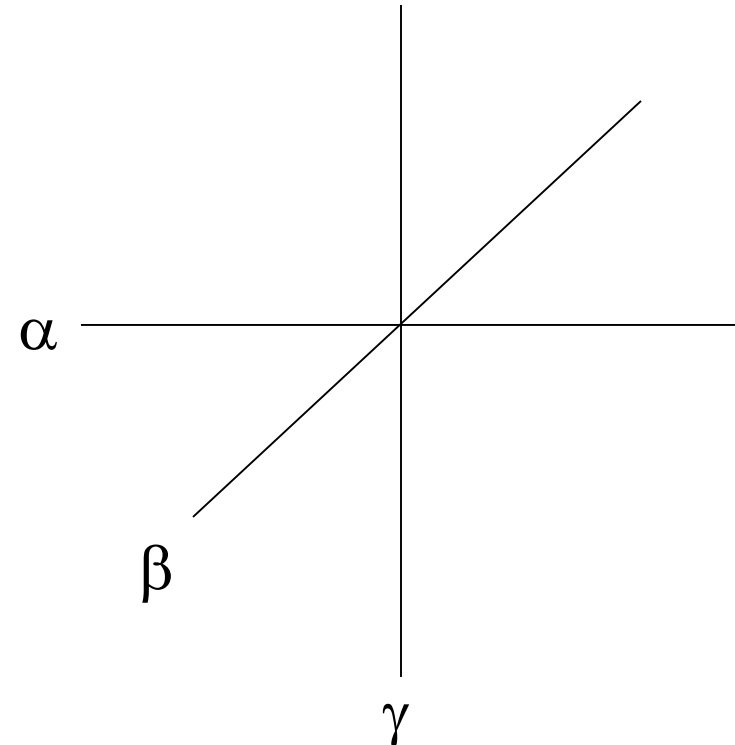
quartz, $\Delta n = 0.009$ (two images with very little separation)

Crystal name	ω	ϵ
rock crystal (quartz)	1.5443	1.5534
calcite	1.6584	1.4864
sapphire	1.768	1.760

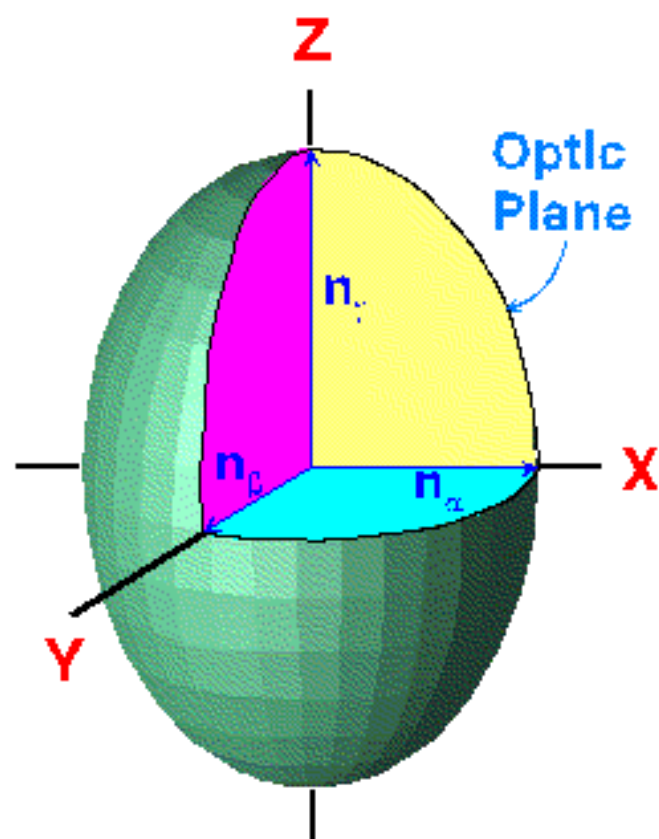
Biaxial Indicatrix 1

- Three refractive indices
- $n_a(\alpha) \neq n_b(\beta) \neq n_c(\gamma)$
- Light travels in different speed depending on directions
- Orthorhombic, monoclinic, and triclinic crystal systems

- by definition, $\gamma > \beta > \alpha$
- $\gamma - \beta > \beta - \alpha$ (+): positive
- $\gamma - \beta < \beta - \alpha$ (-): negative



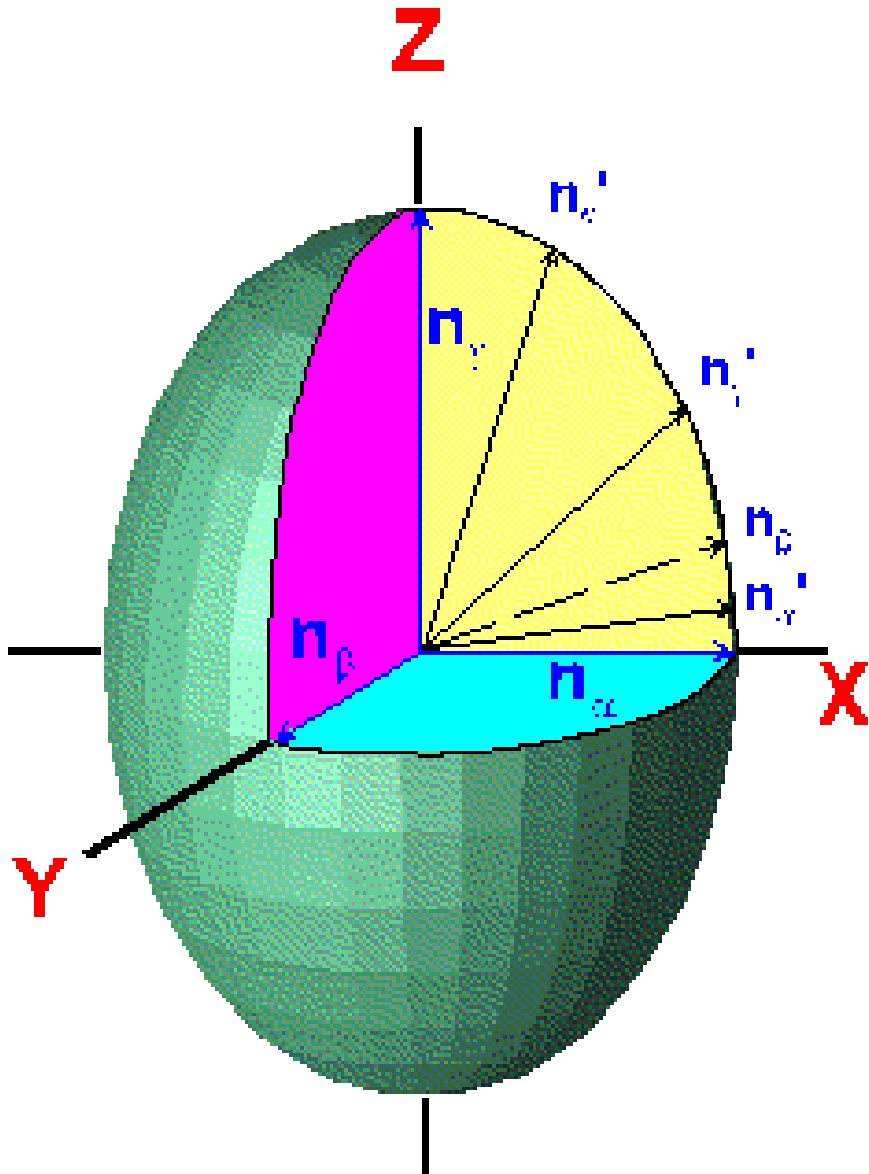
Biaxial Indicatrix 2



Biaxial minerals have three indices of refraction, (n_α , n_β , n_γ) each of which is measured along an indicatrix axis as shown on the left, such that the following relationship holds: $n_\alpha < n_\beta < n_\gamma$.

- XZ** plane with axes n_α and n_γ
- YZ** plane with axes n_β and n_γ
- XY** plane with axes n_α and n_β

Biaxial Crystal: Orientation-Dependence



- γ and β for YZ plan
- γ and α for XZ plan
- α and β for XY plan

Random orientation

- γ , γ' , β , β' , α , and α' depending on the orientation

Biaxial Crystal: Orientation-Dependence

